

# Calculating N<sub>2</sub>O Emissions from the Production of Adipic Acid (Version 2.0)

*Guide to calculation worksheets (December 2007)*

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#### **I. Overview**

##### I.A. Purpose and Domain

This guideline is written for plant managers and site personnel to facilitate measurement and reporting of greenhouse gas direct emissions resulting from adipic acid production. A step-by-step approach is presented to cover every phase of the calculation process from data gathering to reporting. This sector guideline should be applied by the industries whose operations involve the production of adipic acid.

##### I.B. Process Description

The primary use of adipic acid is in the manufacturing of 6,6 nylon. Cyclohexane is used to produce a ketone-alcohol, which is subsequently oxidized with nitric acid to produce adipic acid. Nitrous oxide (N<sub>2</sub>O) is produced as a by-product of this reaction, with an associated emission factor of 300 g N<sub>2</sub>O/kg Adipic acid produced (plus/minus 10 percent depending on the ratio of ketone to alcohol). This emission factor does not incorporate any reductions associated with abatement technologies.

N<sub>2</sub>O emissions can be abated by:

- (1) Treating the off-gases in a reductive furnace.
- (2) Thermally decomposing the N<sub>2</sub>O in a flame reactor (T>900 C) to produce recyclable NO.
- (3) Catalytically decomposing the N<sub>2</sub>O.
- (4) Consuming the N<sub>2</sub>O as an oxidation process feedstock.

The majority of the large adipic acid producers (representing more than 80% of worldwide capacity in 2000) currently treat N<sub>2</sub>O emissions.

##### I.C. Applicability of the tool

Greenhouse gases are also emitted from the fuel combustion process associated with adipic acid production. These emissions are not accounted for in the guidelines described below. Please see the Stationary Combustion guidelines for more details and for the methodology used to estimate these emissions.

## II. Choice of activity data and emission factors

This protocol estimates nitrous oxide emissions based on the quantity of adipic acid produced, an N<sub>2</sub>O emission factor, an abatement technology destruction factor, an abatement technology utilization rate. Adipic acid production and details about abatement technologies should be available from the facility. Default values for an emissions factor and approximate destruction factors for some technologies can be used if factory-specific data is unavailable.

## III. Calculation method used in the tool

The equation used to calculate N<sub>2</sub>O emissions from the production of adipic acid is summarized as follows [source: 2006 IPCC Guidelines]:

$$N_2O \text{ Emissions} = (\text{Adipic Acid Production} \times \text{Emission Factor}) \times (1 - \text{Fraction Abated} \times \text{Utilization Factor})$$

Where:

Adipic Acid Production (metric tons)

Emission Factor (metric tons of N<sub>2</sub>O/metric ton of adipic acid produced)

Fraction Abated (%) – percent of emissions abated by reduction technologies and practices

Utilization Factor (%) – percent of time the abatement technology was in use

Worksheet 1 has been provided to assist in estimating N<sub>2</sub>O emissions from the adipic acid production process. Factory specific data should be used whenever possible. However, default values are provided in Table 1 of Worksheet 1. If a factory specific emission factor is not known, the 2006 IPCC Guidelines provide a default emission factor of 300 g N<sub>2</sub>O /kg adipic acid produced, which converts into 0.3 tonnes of N<sub>2</sub>O /tonne of adipic acid produced. In addition, default Fraction Abated values are given by the 2006 IPCC Guidelines for a variety of abatement technologies.

## IV. Direct N<sub>2</sub>O emissions from Adipic Acid Production

Direct emissions are those that are produced within the boundary (fence) of the reporting facility, or in this case, emissions directly associated with the production of adipic acid.

Worksheet

1. Enter the amount of adipic acid produced (tonnes) during the reporting period in Column A.
2. Enter the N<sub>2</sub>O emissions factor (tonnes of N<sub>2</sub>O /tonnes of adipic acid produced) in Column B. This emission factor should not incorporate any abatement technologies.
3. Calculate N<sub>2</sub>O emissions by multiplying the amount of adipic acid produced by the N<sub>2</sub>O emissions factor (Columns A and B). Column C should now display the product of A and B. If this is not the case, press “F9” to calculate.
- 4. If the facility uses N<sub>2</sub>O abatement technologies, please follow steps 5 through 9. Otherwise, please skip to Step 10.**
5. Enter the type of N<sub>2</sub>O abatement technology in Column A.
6. Enter the percent of N<sub>2</sub>O that is abated through reduction technologies in Column B. These pollution controls must be specific to reducing N<sub>2</sub>O emissions.
7. Enter the utilization factor for the N<sub>2</sub>O abatement technology in terms of the percent of time that technology was used (Column C).
8. Estimate N<sub>2</sub>O emissions by multiplying the potential annual N<sub>2</sub>O emissions calculated in Step 1 above by one minus the N<sub>2</sub>O abatement factor (Column B) times the utilization factor (Column C).
9. Please enter the Global Warming Potential (GWP) for N<sub>2</sub>O. Currently, the GWP for N<sub>2</sub>O over a 100-year time horizon is 310.
10. Calculate Carbon Dioxide Equivalents by multiplying the potential annual N<sub>2</sub>O emissions (Column E) by the GWP of N<sub>2</sub>O (Column F)

## **V. CO<sub>2</sub> Emissions from Fuel Combustion Associated with Adipic Acid Production**

The production of adipic acid consumes various types of fuels. Greenhouse gas emissions associated with this fuel combustion are not directly accounted for in the adipic acid production protocol. Please use the Stationary Combustion guideline to estimate these emissions.

## **VI. References**

**IPCC (2006)**, 2006 IPCC Guidelines for National Greenhouse Gas Inventories. These Guidelines can be accessed at: [http://www.ipcc-nggip.iges.or.jp/public/2006gl/pdf/3\\_Volume3/V3\\_3\\_Ch3\\_Chemical\\_Industry.pdf](http://www.ipcc-nggip.iges.or.jp/public/2006gl/pdf/3_Volume3/V3_3_Ch3_Chemical_Industry.pdf).